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Enantioselective Total Synthesis of (+)-Amphidinolide T1

Arun K. Ghosh* and Chunfeng Liu

Department of Chemistry, University of Illinois at Chicago, 845 West Taylor Street, Chicago, Illinois 60607

Received November 27, 2002; E-mail: arunghos@uic.edu.

Amphidinolides are a series of unique cytotoxic macrolides isolated from the marine dinoflagellates *Amphidinium* sp., and they have shown significant antitumor properties against a variety of NCI tumor cell lines.¹ Amphidinolides are extremely scarce, and as a result, biological studies have been very limited. Amphidinolide T1 (1), a 19-membered macrolide, has shown potent activity against murine lymphoma L1210 as well as human epidermoid carcinoma KB cell lines.² The structure of **1** was initially established by NMR studies. Later, its absolute configuration was elucidated through X-ray analysis by Kobayashi and co-workers.³ The significant clinical potential and unique structural features of the amphidinolide family of natural products have stimulated considerable interest in their syntheses and structure—function studies.⁴ Herein, we report the first total synthesis of (+)-amphidinolide T1.

As outlined in Figure 1, our synthetic strategy for amphidinolide T1 is convergent and involves the assembly of C_1-C_{10} segment 2 and $C_{11}-C_{21}$ segment 3 by an oxocarbenium ion-mediated alkylation and macrolactonization sequence. This alkylation is expected to proceed stereoselectively through a five-membered ring oxocarbenium ion intermediate derived from sulfone 2. Our approach to the synthesis of fragment 2 involves an efficient cross metathesis and hydrogenation sequence between the terminal olefins of 5 and 6 to form the C_4 - C_5 -carbon-carbon bond. Enol ether 4 is designed to be the surrogate of fragment ${\bf 3}$ where the sensitive $C_{16}\mbox{-}exo$ methylene and the C13-hydroxyl group are being protected as a bromohydrin derivative during the Lewis acid-catalyzed alkylation and macrolactonization reactions. We planned to unravel this functionality at the final stage of the synthesis. Two of the three stereocenters in fragment 5 as well as the C_2 and C_3 stereocenters in fragment 4 are accessed by a highly diastereoselective esterderived titanium enolate-mediated syn-aldol reaction.5

The synthesis of C_1-C_{10} segment 2 is outlined in Scheme 1. Aldol condensation of (1R,2S)-1-(N-tosylamino)-2-indanyl propionate (7) with benzyloxypropionaldehyde at -78 °C furnished aldol adduct 8 as a single diastereomer in 90% yield.⁵ Reduction of 8 by lithium aluminum hydride in THF at 23 °C afforded diol 9. This was transformed into γ -lactone 10 by selective sulfonylation of the primary alcohol, displacement of the resulting sulfonate with cyanide, followed by acid-promoted lactonization to afford 10 in 81% overall yield. DIBAL reduction of 10 at -78 °C in CH₂Cl₂ followed by reaction with trimethylsilylethanol and *p*-TsOH afforded acetal 11 and its isomer as a 3.5:1 mixture, which can be separated by column chromatography after removal of the benzyl group by hydrogenolysis. Swern oxidation of the resulting alcohol followed by Wittig olefination furnished alkene 5, one of the substrates for cross metathesis.

Cross metathesis⁶ of **5** with oxazolidinone derivative **6**⁷ in the presence of 5 mol % second generation Grubbs's catalyst⁹ afforded a 1:1 mixture (*E:Z*) of cross metathesis product **12** in 60% yield along with alkene dimers of **5** and **6** which can be separated by silica gel chromatography. Exposure of these alkene dimers to 5 mol % second generation Grubbs's catalyst further afforded cross

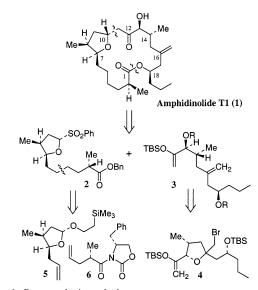
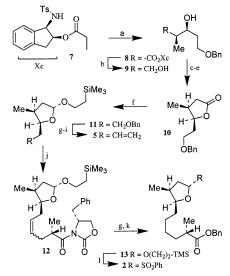
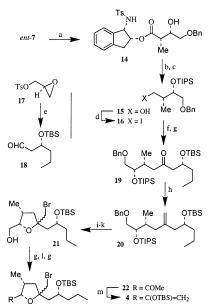


Figure 1. Retrosynthetic analysis. Scheme 1^a



^aReaction conditions: (a) TiCl₄, *i*-Pr₂NEt; BnO(CH₂)₂CHO (90%); (b) LAH, THF, 23 °C (91%); (c) PhLi, TrisCl, THF, -78 to 23 °C (86%); (d) NaCN, DMSO, 80 °C (95%); (e) aq. HCl, MeOH, 23 °C (98%); (f) Dibal-H, -78 °C, then TMS(CH₂)₂OH, *p*-TsOH, MgSO₄, 23 °C (91%); (g) H₂, Pd/C, EtOAc (98%); (h) Swern Ox; (i) Ph₃P=CH₂, THF (84%): (j) **6**, CH₂Cl₂, 40 °C Cl₂(PCy₃)(IMes)Ru=CHPh (96%); (k) BnOH, BuLi, THF, 0 °C (85%); (l) HSO₂Ph, CaCl₂, CH₂Cl₂, 23 °C (95%).

metathesis product **12** in 36% yield (1:1 mixture of *E:Z* olefins). Thus, the overall yield of cross metathesis product **12** was 96% after two cycles. Catalytic hydrogenation of the alkene mixture followed by treatment of the saturated derivative with lithium phenylmethoxide furnished benzyl ester **13** in 85% yield. Exposure of **13** to phenylsulfinic acid and CaCl₂ afforded sulfone derivative **2** and its isomer as a 7:1 mixture in 95% yield.⁸

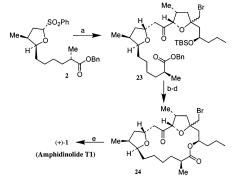


^a Reaction conditions: (a) TiCl₄, *i*-Pr₂NEt; BnOCH₂CHO (93%); (b) TIPSOTf, *i*-Pr₂NEt, 0 °C (95%); (c) Dibal-H, -40 °C (91%); (d) I₂, PPh₃, Imidazole, 23 °C (95%); (e) i. BuLi, 1,3-dithiane; ii. EtMgBr, CuI (64%, two steps); iii. TBSOTf, i-Pr2NEt; iv. MeI, CaCO3, aq. MeCN (73%, two steps); (f) t-BuLi, Et₂O, -78 to 23 °C, then 18 (80%); (g) TPAP, NMO, 23 °C; (h) Cp₂TiMe₂, THF, 80 °C (84%); (i) Li/NH₃, -33 °C, THF; (j) LAH, THF, 23 °C (90%); (k) NBS, 0 to 23 °C (91%); (l) MeMgBr, -78 to 23 °C; (m) LiHMDS, TBSCl, HMPA, -78 to 23 °C (95%).

Synthesis of the C11-C22 fragment is outlined in Scheme 2. Aldol reaction of ent-7 with benzyloxyacetaldehyde afforded a single syndiastereomer (14) in 95% yield.⁵ Protection of the resulting alcohol as a TIPS ether followed by Dibal-H reduction furnished alcohol 15 which was readily converted into the corresponding iodide (16). Reaction of glycidyl tosylate 17 with lithium 1,3-dithiane followed by ethyl cuprate provided the corresponding secondary alcohol which, after protection as a TBS ether and removal of dithiane, afforded aldehyde 18.9 Treatment of iodide 16 with t-BuLi generated the corresponding alkanyllithium, which was reacted with aldehyde 18 to afford a 1:1 diastereomeric mixture of alcohols which, upon TPAP oxidation, provided ketone 19.10 Olefination of 19 utilizing Petasis conditions¹¹ furnished alkene 20. Reductive removal of the benzyl and TIPS ethers¹² followed by reaction of the resulting diol with NBS afforded bromotetrahydrofuran 21 as a 3:1 diastereomeric mixture. Our motive for forming this bromolactone is to protect the C13-alcohol as well as to protect the sensitive exomethylene group during the oxocarbenium ion-mediated alkylation process.¹³ The hydroxymethyl group of **21** was then converted to methyl ketone 22 in 60% overall yield. Treatment of 22 with LiHMDS followed by reaction of the resulting enolate with TBSCl furnished the vinyl ether segment (4).

Our subsequent synthetic strategy calls for the assembly of fragments 2 and 4 by an oxocarbenium ion-mediated alkylation reaction; however, this proved to be a formidable task. Our successful path using modified Ley's protocol¹⁴ is presented here.¹⁵ Treatment of 2 and 4 in the presence of 2 equiv of $AlCl_3$ at -35°C resulted in only very low yield (10%) of desired product 23. Under optimized conditions in the presence of excess AlCl₃ (6 equiv) and DTBMP (1.2 equiv) coupling product 23 was obtained in 73% yield as a single isomer (by 1H- and 13C NMR analysis).16 The depicted trans-stereochemistry is assigned on the basis of analysis of NOESY data. The C18-silyl ether was then removed by exposure to HF·Py, and subsequent hydrogenolysis removed the benzylester. Macrolactonization of the resulting hydroxy acid under

Scheme 3^a



^{*a*} Reaction conditions: (a) **4**, DTBMP, AlCl₃, -35 °C (73%); (b) HF•Py, Py (87%); (c) H₂, 10% Pd-C; (d) 2,4,6-(Cl₃)PhCOCl, *i*-Pr₂NEt, then DMAP, toluene (71%); (e) Zn, NH₄Cl, EtOH, 80 °C (61%).

Yamaguchi conditions¹⁷ afforded macrolactone 24 in 71% yield (Scheme 3). Reductive unmasking of the bromoether with Zn and NH₄Cl in ethanol provided synthetic amphidinolide T1 (1). Spectral data (¹H and ¹³C NMR) of synthetic 1 ($[\alpha]^{23}_{D}$ +16, c 0.1, CHCl₃) is in agreement to that of natural amphidinolide T1 (lit.³ $[\alpha]_D^{23}$ $+18, c 0.3, CHCl_3$).

Thus, a stereocontrolled and convergent synthesis of (+)amphidinolide T1 has been achieved. The functional group tolerabilty of the oxocarbenium ion-mediated alkylation reaction along with efficient cross methathesis, diastereoselective aldol reactions, and use of a cyclic bromoether as a novel protection of the exomethylene group are noteworthy features.

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Supporting Information Available: Experimental procedures and spectral data for compounds and ¹³C NMR for compounds 2, 4-24 and 1 (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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